A laboratory volcano (I): application to Kilauea

I) Fire fountain:

kinetic energy is transferred to potential energy (ballistic approximation):

velocity =
$$(2 g H)^{1/2}$$

H: fire fountain height g: acceleration of gravity

$$H = 400 \text{ m}$$



velocity = 100 m/s



Fire fountain during Puu o'O eruption (Kilauea, Hawaii)

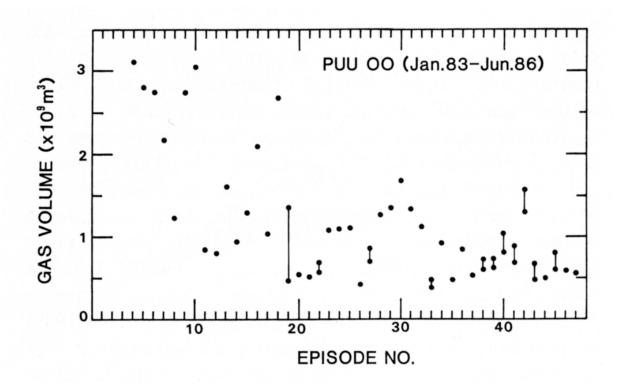
gas flux at vent = velocity x surface = gas volume / time

gas volume = velocity x surface x time

A laboratory volcano (2): application to Kilauea

gas volume at the vent is determined from fire fountain height

Perfect gas law: P V = constant P : pressure; V: gas volume $P_g vent V_g vent = P_g reservoir V_g reservoir$



₹ gas volume in magma chamber

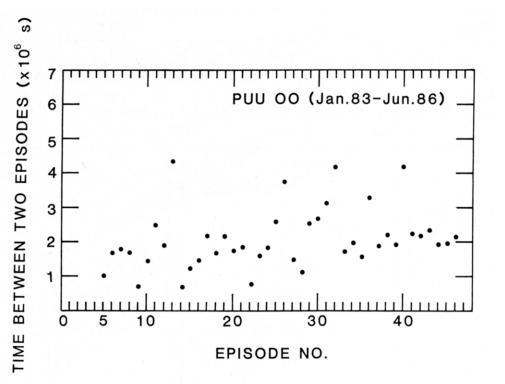
decrease in gas volume over the 3 years and a half

A laboratory volcano (3): application to Kilauea

2) Gas flux in magma chamber Q:

gas volume in magma chamber from fire fountains height

Intermittency between fire fountain: roughly constant





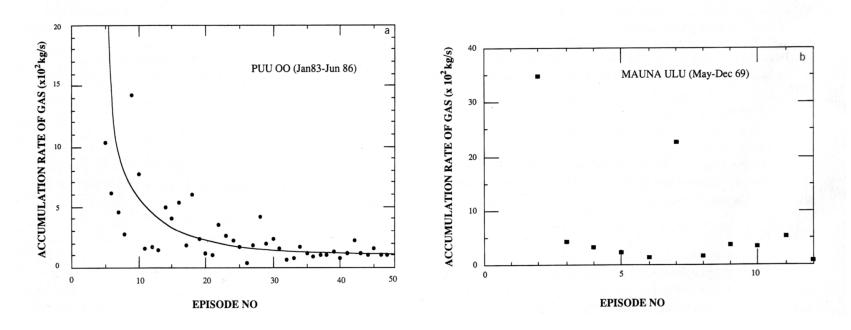
Gas flux in magma chamber is a gas volume at depth over a time during which it has been accumulated

$$Q = \frac{\text{gas volume (2)}}{\text{intermittency (1-2)}}$$

A laboratory volcano (4): application to Kilauea

Gas flux in magma chamber Q:

for Kilauea volcano (Puu O'o and Mauna Ulu eruptions) calculated for a reservoir at 4 km depth and from fire fountain height



decrease in gas flux in reservoir in time

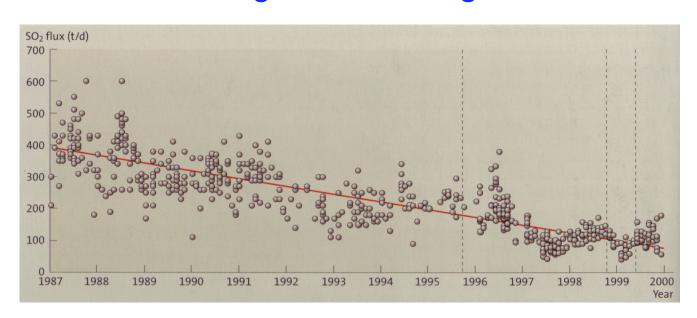


A laboratory volcano (5): application to Kilauea

The critical gas flux in reservoir, determined from fire fountain height, is approximatively 50 kg/s at Kilauea volcano

 SO_2 is a magmatic gas which forms at large depth, which may be indicative of the relative evolution of gas volume and gas flux

SO₂ is easily measured at the vent (mass spectometer: Cospec, FTIR...)

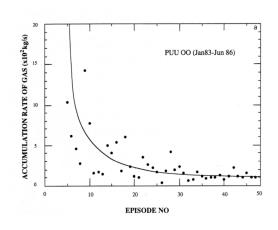


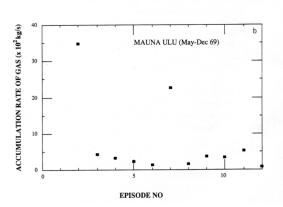
decrease in gas flux during fire fountains is confirmed for the period after the fire fountain episodes (effusive activity)

A laboratory volcano (6): application to Kilauea

Critical gas flux:

estimated from the fire fountain height at transition fire fountain/ effusive





Q_c: critical gas flux: 50 kg/s

$$Q_{\mathrm{c}} = B \frac{\sigma^4}{
ho_{\mathrm{liq}}^3 \mu_{\mathrm{liq}} d^4}$$
 B = f(\varepsilon, \mathrm{r}_{\mathrm{c}}, \mathrm{r}_{\mathrm{t}})

 \rightarrow d = f(critical gas flux)

 Q_c : critical gas flux (m³/s) σ : surface tension (kg.s-²) ρ_{liq} : liquid density (kg.m-³) μ_{liq} : liquid viscosity (Pa.s) d: bubble diameter (m)

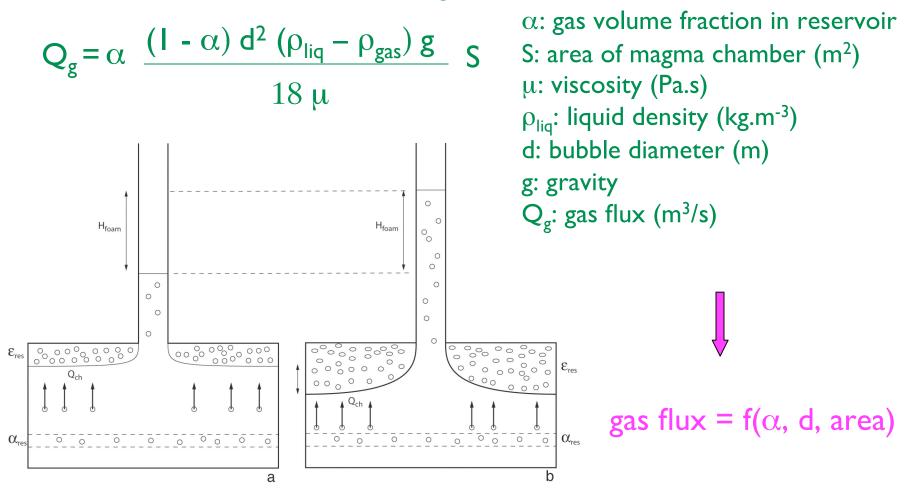
 ϵ : gas volume fraction r_t : reservoir radius (m)

r_c: conduit radius (m)

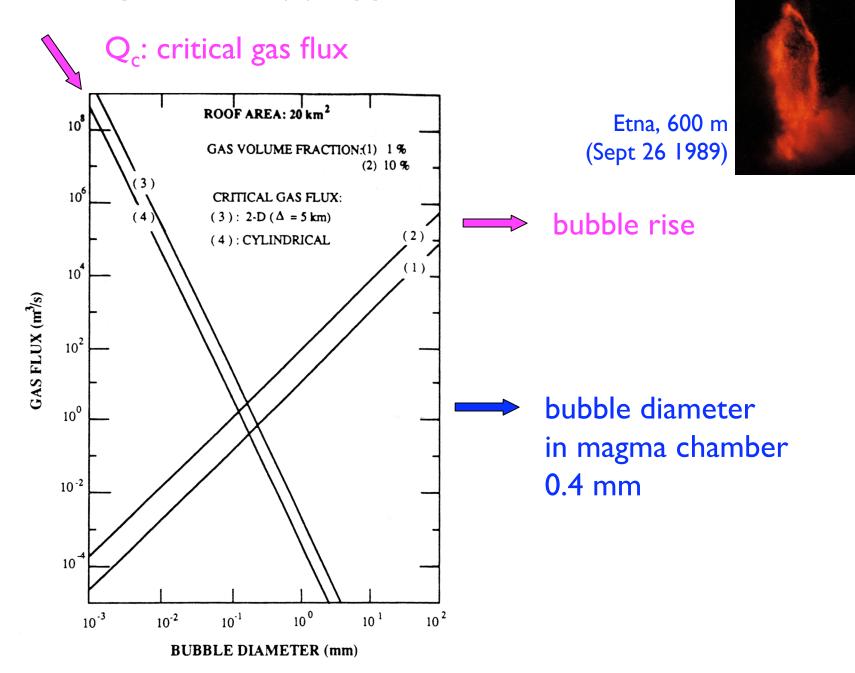
A laboratory volcano (7): application to Kilauea

Gas flux in magma chamber Qg:

from bubble rise velocity $v_b : Q_g = \alpha v_b S$

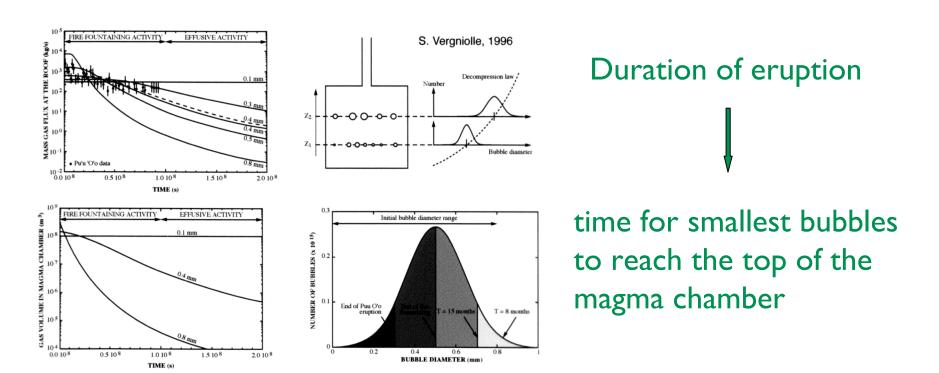


A laboratory volcano (8): application to Kilauea



A laboratory volcano (9): application to Kilauea

Duration of an eruption: evolution of a polydisperse suspension of bubbles initially widespread everywhere in a closed magma reservoir



Eruption stops when magma chamber has been depleted of most of its bubble content

A laboratory volcano (10): application to Kilauea

General trend: decrease of gas flux in time

Ist bubbles are the largest and the latest bubbles the smallest

2 dimensionless times based on observed durations : $\tau_{\rm f}^*$ and $\tau_{\rm e}^*$

Duration of fire fountain τ_f :

$$\tau_{f}^{*} = \tau_{f} \frac{d_{0}^{2} (\rho_{liq} - \rho_{gas}) g}{18 \mu h_{c}}$$
 $\tau_{f}^{*} = 1.1 : Puu O'o eruption (1987-now)$

$$\tau_{f}^{*} = 1.3 : Mauna Ulu eruption (1969-1971)$$

Duration of eruption τ_e :

$$\tau_{e}^{*} = \tau_{e} \frac{d_{0}^{2} (\rho_{liq} - \rho_{gas}) g}{18 \mu h_{c}}$$

$$\tau_{e}^{*} = 4.2 : Puu O'o eruption (1987-now)$$

$$\tau_{e}^{*} = 4.3 : Mauna Ulu eruption (1969-1971)$$

Mean bubble diameter: 0.4 mm: Puu O'o and Mauna Ulu eruptions

Degassing layer = 2 km (Puu O'o) and 300 - 500 m (Mauna Ulu)

Duration of Puu O'o eruption is 6 times longer than Mauna Ulu

A laboratory volcano (11): application to Kilauea

Lava pond, Puu O'o eruption (Kilauea, March 1988)

Makaopuhi lava lake, 1965: height = 83 m = h_c viscosity = 50 Pa.s = μ



Temperature measurements + model of thermal convection

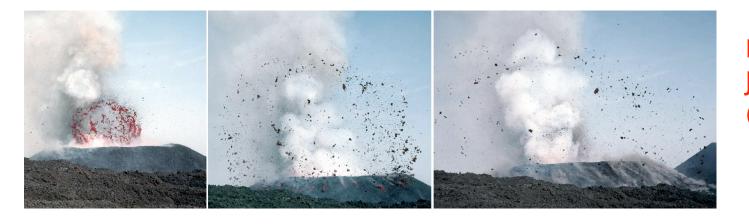
for the first 5 months of emplacement:
no thermal convection because of bubble rise

$$\tau_e^*$$
 = 5 months
$$\tau_e^*$$
 = 4.2 mean bubble diameter d_0 = 1 mm

same order of magnitude as for Puu O'o and Mauna Ulu eruptions

A laboratory volcano (12): application to Etna (Italy)

Recent activity: a series of regular eruptive episodes (64 episodes in 2000; 15 episodes in 2001; id 1989; 1998) each episode: a series of Strombolian explosions a episode lasts a few hours, intermittency: a few days



Etna, July 2001 (Pfeiffer)

Each explosion corresponds to large bubble bursting (diameter of several meters)

Acoustic measurements



gas volume at vent

A laboratory volcano (13): application to Etna (Italy)

FTIR measurements: gas composition with high CO₂/S and S/Cl violent emptying of a foam layer accumulated at 1.5-2 km depth

Perfect gas law Gas volume at depth of reservoir

+ intermittency between episodes (3-5 days)

Gas flux at depth of reservoir Bubble diameter: 0.7 mm

9 May-13 July 2001: trachybasalt at summit (South East Crater)

I7 July-9 August 2001: flank-eruption: alkali-rich basalt (primitive):
3.4 wt% H₂O; 0.11-0.41 wt% CO₂.

Origin of flank eruptions is not well-known but in July 2001 associated with new fresh magma

Etna, flank eruption 2001



Basaltic magma reservoirs

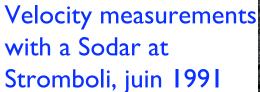
- I. Physical parameters
- 2. A laboratory volcano:



3. Gas volume measurements



4. Conclusion





Basaltic magma reservoirs

- I. Physical parameters
- 2. A laboratory volcano:
- 3. Gas volume measurements



a) ballistics and radar



b) acoustic records

4. Conclusion

Velocity measurements with a Sodar at Stromboli, juin 1991



Direct measurements (I): ballistics

Strombolian explosion:

A series of bubbles bursting at the top of the magma column

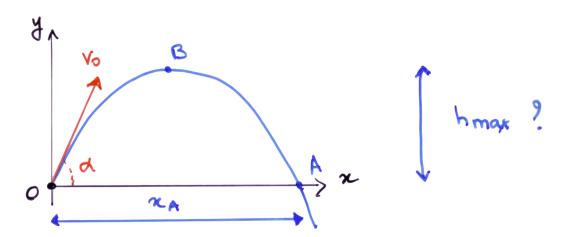
duration: a few seconds

Intermittency: minutes to hours



Stromboli, 1992

Ballistic of a point:



assumption: no air resistance

$$h_{\text{max}} = \frac{v_0^2 \sin^2(\alpha)}{2 g}$$

Direct measurements (2): ballistics

For a vertical gas jet $v_0 = (2 g h_{max})^{1/2}$

(kinetic energy is transformed into potential energy)

Example: Stromboli

$$h_{max} = 100 \text{ m}$$
 $v_0 = 50 \text{ m/s}$

 v_0 is not too sensitive to h_{max} , which is difficult to measure

Impact on the ground: x_a

$$x_a = \frac{v_0^2 \sin(2\alpha)}{2 g}$$
 $x_a \text{ is maximum for } \alpha = 45^\circ$
 $\alpha = 60^\circ; v_0 = 50 \text{ m/s}; x_a = 220 \text{ m}$

Maximum distance reached by ballistics ejecta: a few km. If a small plume is formed, small ejecta can be carried further away

Direct measurements (3): ballistics

Bubble rise in a magma:

```
\mu\text{:} viscosity (Pa.s); \rho_{\text{liq}} and \rho_{\text{gas}} liquid and gas density (kg/m³);
```

$$v_{gas} = \frac{d^2 (\rho_{liq} - \rho_{gas}) g}{18 \mu}$$

balance between viscous resistance in magma and bubble buoyancy

Basalt: a bubble of I mm in diameter (d) takes 6 months to rise I km

particle in gas jet:

$$v_{part}^2 = \frac{2 \pi d (\rho_{liq} - \rho_{gas}) g}{3 C_d \rho_{air}}$$

 C_d : drag coefficient (turbulent) = 1.18

balance between friction in air and particle weight

Remark: Stromboli; crystal settling: d = 1 mm; $v_{cryst} = 56 \text{ m/s}$

Apparent velocity:
$$w = v_{gas} - v_{part}$$
ballistics $v_{part} = 50 \text{ m/s}$: particle velocity
$$V_{gas} = 100 \text{ m/s}$$
: gas velocity

Direct measurements (4): ballistics

Particle diameter (cm) Apparent Particle velocity ERUPTION I 2594 EVENTS 2562 EVENTS PLOTTED ERUPTION I 2594 EVENTS €00. 2594 EVENTS PLOTTED 32 OVERFLOW ALL EVENTS, AVERAGE 2.20 CM FRE GUENCY 300-ALL EVENTS. AVERAGE : 26.16 M/SEC QUENCY FRE 200 20 100-10 20 28 32 0 20 25 30 35 40 45 PARTICLE SIZE (CM) APPARENT PARTICLE VELOCITY (M/SEC) Stromboli, 32: 28. ERUPTION 2:152 EVENTS 1971 152 EVENTS PLOTTED 24. ALL EVENTS, AVERAGE: 2.48 CM 20. FREQUENCY ERUPTION 2:152 EVENTS : 6 152 EVENTS PLOTTED FREQUENCY ALL EVENTS. AVERAGE: 15.28 M/SEC

20

30 35 40

APPARENT PARTICLE VELOCITY (M/SEC)

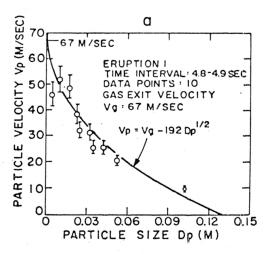
Fig. 9. Frequency distributions of particle sizes and apparent particle velocities for the bulk of the two eruptions studied.

32

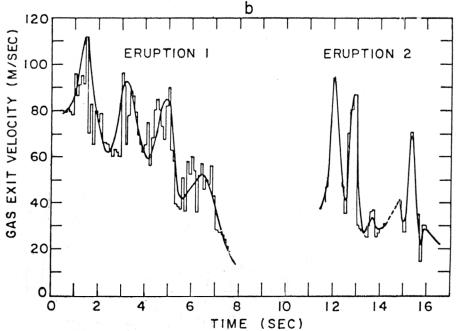
PAPTICLE SIZE (CM)

Direct measurements (5): ballistics

Stromboli, 1971

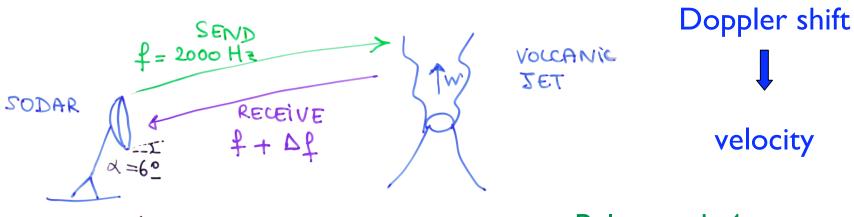


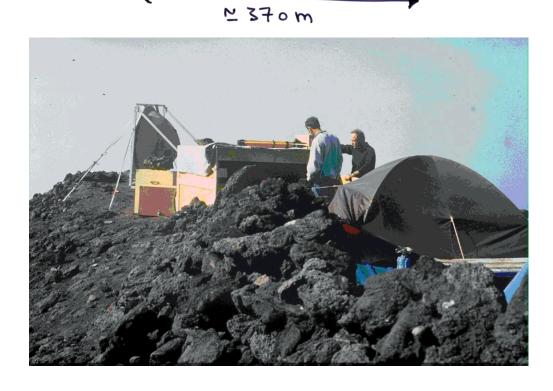
Apparent particle velocity: the smallest particle moves at the same velocity as the gas



Oscillations in gas jet: attributed to resonances of the conduit (300 m length)

Direct measurements (6): radar: Acoustic sounder





Pulse: each 4 s; τ_{on} = 50 ms

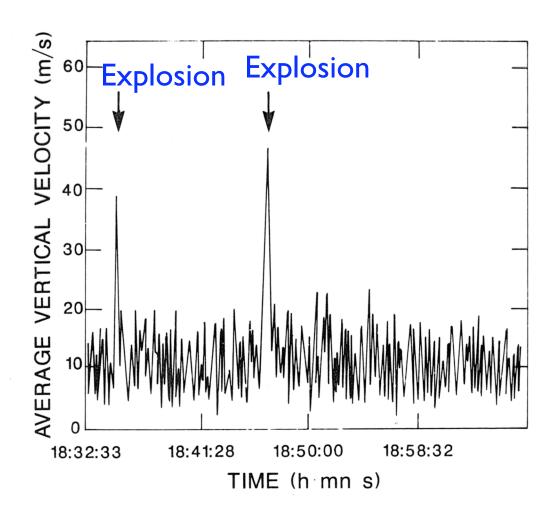
 $\Delta f / f = -2 w \sin(\alpha) / c$

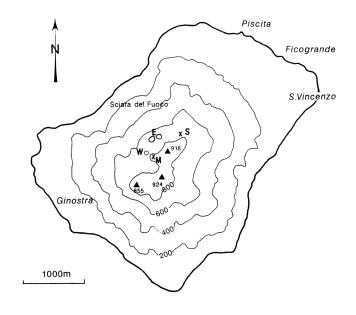
c: sound speed

air: 340 m/s

hot gas : 700 m/s

Direct measurements (7): Acoustic sounder

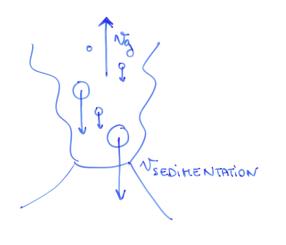






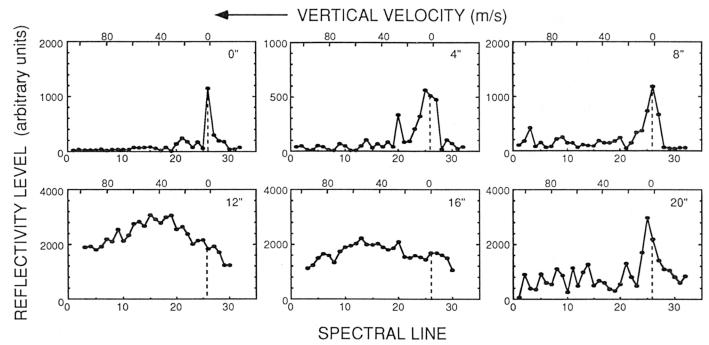
Summit vents at Stromboli

Direct measurements (8): Acoustic sounder



$$w = v_g - v_{part}$$

A range of particles velocity because of a range of particle diameter



Basaltic magma reservoirs

- I. Physical parameters
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 - a) ballistics and radar
 - b) acoustic records
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Acoustic measurements at Yasur, Vanuatu, 1993





Direct measurements (9): Acoustic records



Yasur, 1994



Microphone, Yasur, 1994

Sound is produced by a volcano has a strong intensity

Sound: variation of pressure, which propagate in a compressible fluid correspond to density variations in an elastic media radiation of sound in air from volcano vent

Sound waves are longitudinal waves (P): elements of fluid move parrallel to the direction of propagation move back and forth: zones of compression and dilatation

Acoustic: related to oscillatory motion

identical to seismology except propagation in air

Direct measurements (9): Acoustic records



Microbarograph at Yasur (oct 2002)



source = gas

Yasur, (oct 1993)

Sound is produced by a volcano Informations on volcanic activity recording sound waves is a way to monitor volcanic activity

What is the source of the sound?

Frequency = size of the source

Amplitude = overpressure ejecta velocity

Phase = initial condition

Evolution in time of the eruption characteristics

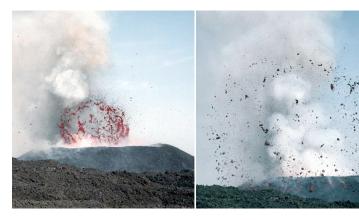
Direct measurements (10): Acoustic records

First developped on Strombolian explosions: permanent activity and not dangerous (close to the vent)

A series of bubbles bursting at the top of the magma column duration: a few seconds

Intermittency: minutes to hours

Etna, July 2001 (Pfeiffer)





Ejecta velocity = 50 m/s

Gas velocity = 100 m/s

Ejecta size = 2 cm (Stromboli)

= 20 cm (Etna)

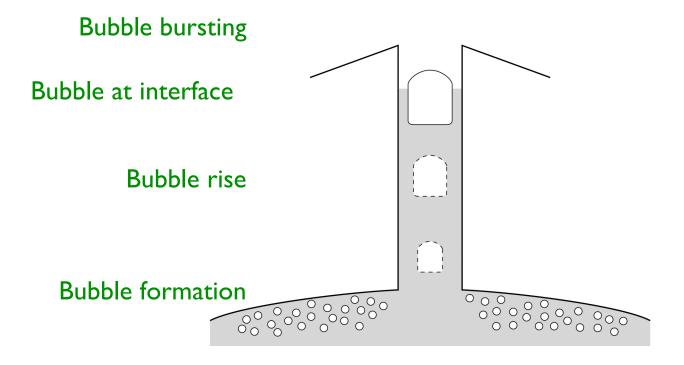
Buble diameter = several meters (Heimaey, Etna, Erebus, Kilauea...)

Direct measurements (11): Acoustic records

Etna, July 2001

Gas exists in conduit for basaltic eruption:
strombolian explosions: a few meters long
fire fountains: several tens of meters long





sound waves

bubble size and pressure

bubble pressure at depth



Eruption dynamics

Direct measurements (12): Acoustic records

Strombolian explosions or fire fountains: origin of large bubbles at depth (carry informations from depth) understanding large bubbles, is understanding eruptive behaviour



Fire fountain
Puu o'O eruption
(Kilauea, Hawaii)

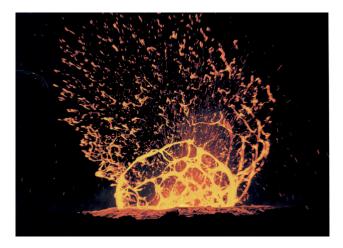


Stromboli, 1992

Acoustic measurements: remote measurements of « magmatic » bubbles at the vent

Observations at the surface + flow model in conduit quantitative information at depth

Direct measurements (13): Acoustic records



Bubble bursting at the surface of lava lake (Kilauea, Hawaii)

Sound is produced by bubble breaking at the surface of lava column / lava lake



bubble must have a residual overpressure

Bubble breaking = balloon bursting?

balloon bursting sound speed = 340 m/s + radius = 2 m frequency = sound speed / radius

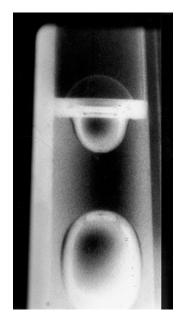
frequency = 150 Hz

Frequency measured on volcanoes: I - I0 Hz

more than I order of magnitude discrepancy the mechanism is not bubble bursting such as for a balloon

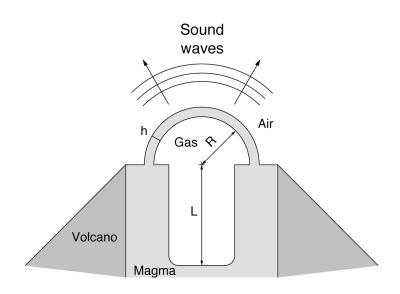
Direct measurements (14): Acoustic records

Bubble vibration model: bubble arrives at the surface of lava column with an overpressure ΔP



Laboratory experiments (silicone oil 12.5 Pa.s, tube diameter = 0.14 m)

Sketch of bubble vibration model

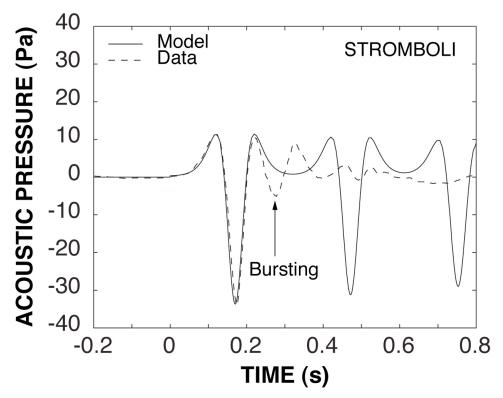


Bubble grows and pass equilibrium position with non-zero velocity: bubble overshoot, so the bubble become larger than equilibrium has a pression less than equilibrium. Gas compressibility acts to restore the equilibrium pressure, so bubble shrinks.



Direct measurements (15): Acoustic records

Bubble vibration model: comparison between theory and measurements: best fit method





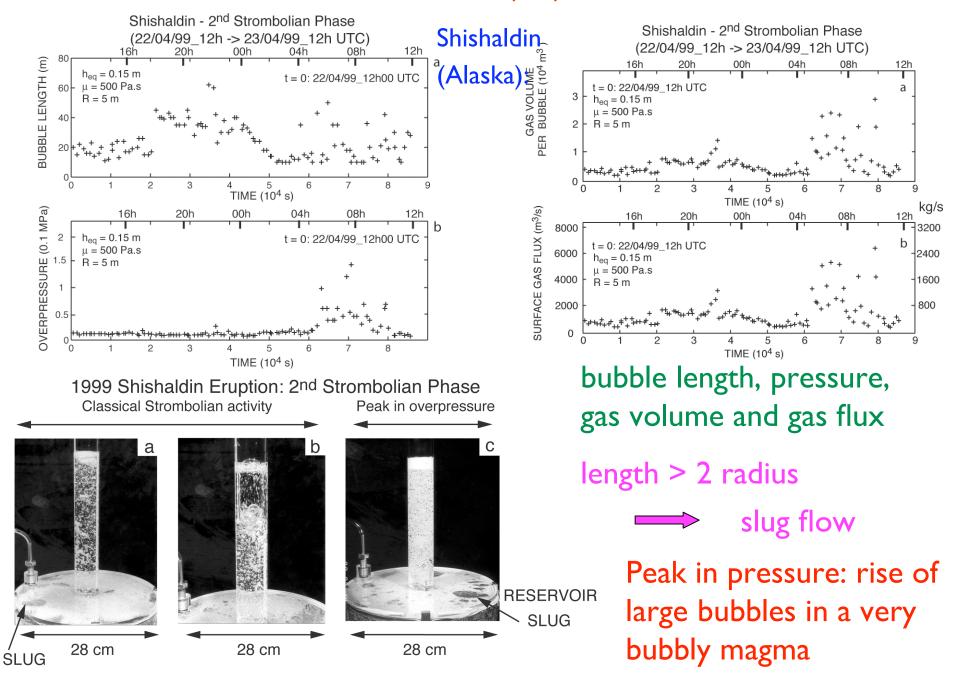
Bubble bursting at the surface of lava lake (Kilauea)



Model for source of sound: bubble radius, length, overpressure and their evolution in time

Model is very robust for gas volume estimates because it is based on frequency content

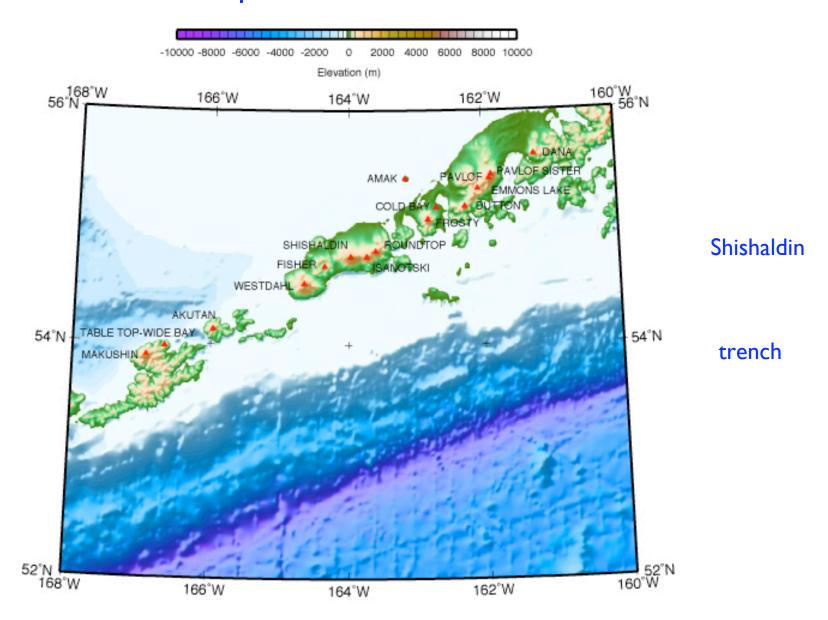
Direct measurements (16): Acoustic records



Direct measurements (17): Acoustic records

The 1999 Shishaldin eruption

Alaska subduction zone



Direct measurements (18): Acoustic records

Shishaldin, Alaska, USA

Formation of a basaltic plume height > 16 km

30 % of volcanoes in subduction zone are basaltic (basaltic andesite)

Basaltic plume is very rare

Eruption finishes with Strombolian activity (a series of large bubbles breaking)

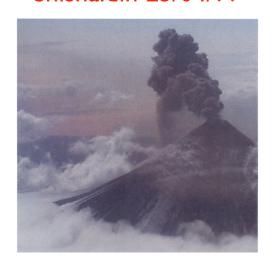
Plume = very analogous to explosive eruptions

Acoustic measurements origin of basaltic Subplinian activity transition Subplinian- Strombolian





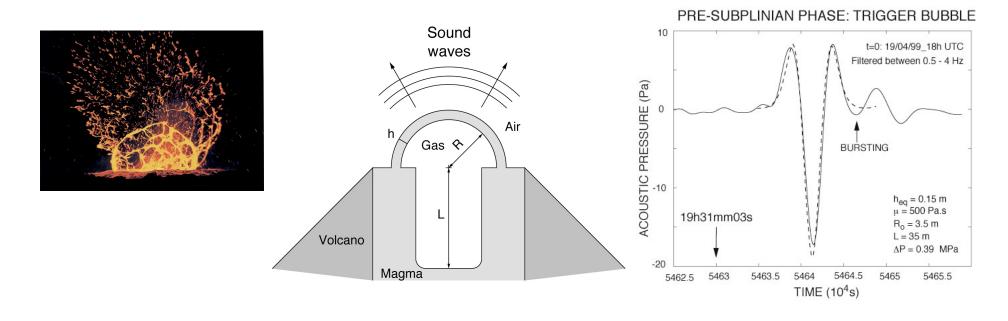
Shishaldin 23/04/99



Direct measurements (19): Acoustic records

Transition to the Subplinian phase:

Very different acoustic pressure than during plume: identical to a single strombolian explosion = bursting of a large overpressurised bubble formed at the depth of the reservoir



Model of bubble vibration:



 $L = 35 \text{ m}; \Delta P = 0.39 \text{ MPa}$

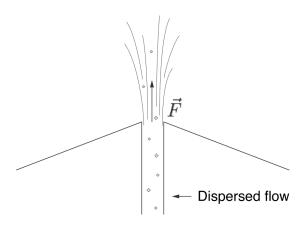
Direct measurements (20): Acoustic records

Different types for source of sound:

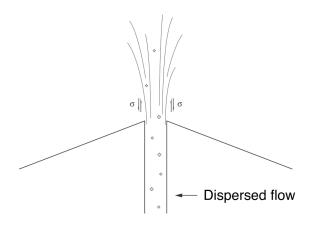
monopole source

Sound waves Gas ❖ Volcano

dipole source



quadrupole



variation of mass flux

+ variation of external force (walls of conduit)

+ variation of stress (turbulence)

Acoustic power Π_m is proportional to (U: gas velocity)ⁿ

$$\begin{split} \Pi_{\rm m} &= K_{\rm m} \frac{4\pi R_{\rm b}^2 \rho_{\rm air} U^4}{c} & \Pi_{\rm d} = \frac{K_d \, \rho_{\rm air} \, A_d \, U^6}{c^3} \\ {\rm K_m = I \; (sphere)} & {\rm K_d = I/3} \\ {\rm (I/I6 \; flat \; circular)} & {\rm (or \; 0.013)} \end{split}$$

$$\Pi_{\rm d} = \frac{K_d \, \rho_{\rm air} \, A_d \, U^6}{c^3}$$

$$K_{\rm d} = 1/3$$
 (or 0.013)

$$\Pi_{\rm q} = rac{K_q
ho_{
m air} \pi R_c^2 U^8}{c^5}$$
 $K_{
m q}$ = 3 10⁻⁵ - 10⁻⁴

 K_m : empirical constant; R_b and R_c : bubble and conduit radius; c: sound speed

Direct measurements (21): Acoustic records

Subplinian phase:

Complex waveform





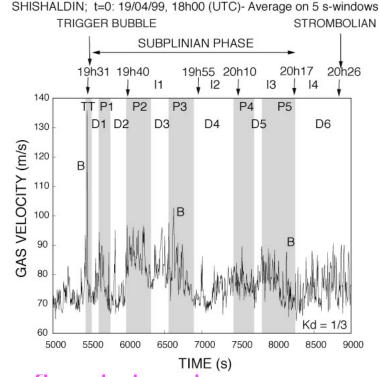


$$\Pi = \frac{\pi r^2}{\rho_{\rm air} cT} \int_0^T |p_{\rm ac} - p_{\rm air}|^2 dt$$

$$\Pi_{\rm d} = \frac{K_d \, \rho_{\rm air} \, A_d \, U^6}{c^3} \quad \mathbf{K_d} = \mathbf{I/3}$$

6 periods

gas flux = velocity x area



In very good agreement with the gas flux deduced from the height in the atmosphere

gas flux = gas volume/time



gas volume= 1.5 10⁷ m³

Direct measurements (22): Acoustic records

Other basaltic volcanoes have magmatic plume eruptions:



Cerro Negro (Nicaragua)



Lopevi (Vanuatu)

1992, similar to 1968 eruption (6 wt% $H_2O + I$ wt% CO_2);

10 km high plume (2001, 2003, next?)

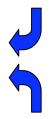
Similar mechanisms than Shishaldin?

Basaltic magma reservoirs

- I. Physical parameters
- 2. A laboratory volcano:
- 3. Gas volume measurements



4. Conclusion



Velocity measurements with a Sodar at Stromboli, juin 1991



Conclusion (I):

Hawaiian eruption: cyclicity between fire fountains and effusive activity



Lava flow into sea (Kilauea, March 1988)



Fire fountains (Puu O'o eruption)

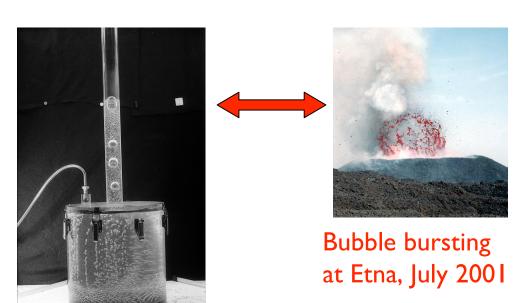
Cyclicity is related to the degassing reservoir: formation of a foam at the top of the reservoir

bubble diameter in reservoir = 0.4 mm

Eruption stops when magma chamber does not contain much bubbles

Conclusion (2):

Strombolian eruption: a series of explosions



Alternance between gas pockets and no activity: is similar to Hawaii but corresponds to a larger magma viscosity

Etna: cyclic fire fountains + gas composition gives unambiguous evidence for being driven by gas accumulation at depth

Eruption dynamics is driven by gas



Need for measurements of gas volume such as radar and acoustic measurements, ...