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Magmatic volatile contents and degassing-induced crystallization at Volcán Jorullo, Mexico: Implications for melt evolution and the plumbing systems of monogenetic volcanoes

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ABSTRACT

Monogenetic basaltic cinder cones are abundant on Earth and exhibit a wide range of eruptive styles, including violent explosions. However, the mechanisms driving explosive cinder cone eruptions are still poorly understood. Here we investigate relations between volatiles, degassing, and crystallization in a longlived, historical, cinder cone eruption to better understand the plumbing systems of monogenetic volcanoes. We present volatile (H₂O, CO₂, S, Cl), major, and trace element data for olivine-hosted melt inclusions, estimates of olivine residence times based on Fe-Mg zoning, and measurements of groundmass crystallinity for tephra from the eruption of Volcan Jorullo, Mexico. Jorullo melt inclusions trapped some of the most volatile-rich (≤5.3 wt.% H₂O, ≤1000 ppm CO₂), primitive (≤10.5 wt.% MgO) melts yet measured in an arc setting, as well as more degassed, evolved compositions. Furthermore, the melt inclusions record temporal changes in both melt composition and crystallization. Early erupted inclusions are Mg-rich and record variable trapping pressures (10-400 MPa), whereas late inclusions were trapped only shallowly beneath the volcano (3-19 MPa) and contain increasingly evolved melts. Disparities between the compositions of the melt inclusions and the whole-rock lava samples provide evidence for a two-stage crystallization process: 1) cooling-induced fractionation of amphibole+olivine±clinopyroxene in the lower crust, which drove the bulk melt evolution over time; and 2) degassing-induced crystallization of melts during ascent at pressures <400 MPa. Additionally, olivine residence times calculated from diffusion profiles suggest that as the eruption progressed, olivine crystals were being stored for longer periods of time (up to 1300 days) within more evolved melts that had risen from depth. These data, taken together with temporal decreases in crystallization depths and increases in groundmass crystallinity, suggest the formation of a shallow reservoir (or simply an enlarged region of the conduit) beneath the volcano late in the eruption. Shallow storage permitted degassing and crystallization of the Jorullo melts, and facilitated assimilation of the host rock by the resident magma.

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36 1. Introduction

Water is fundamental to magma production in subduction zones 39 by initiating melting of the mantle wedge above the downgoing slab. 40 Furthermore, the presence of water and other volatiles in magma prior 41 to eruption affects crystallization during ascent and eruption 42explosivity. Melt inclusion analyses have shown that arc magmas 43 44 can have high H₂O (3.5-8 wt.%; Sisson and Layne, 1993; Roggensack 45 et al., 1997; Cervantes and Wallace, 2003; Wade et al., 2006; Spilliaert et al., 2006) that may correlate with high explosivity (Roggensack 46 47et al., 1997; Spilliaert et al., 2006). Degassing of these H₂O-rich magmas during ascent may also induce rapid and extensive crystal-48

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lization (e.g., Sisson and Layne, 1993; Métrich et al., 2001; Roggensack, 49 2001; Atlas et al., 2006; Blundy et al., 2006). However, most studies 50 have focused on persistently active, larger volcanoes, often with 51 evolved melt compositions.

Our research focuses on high-Mg basaltic magmas from a cinder 53 cone eruption. Cinder cones are the most numerous volcanoes on land 54 (Vespermann and Schmincke, 2002), and due to their small volumes 55 and relatively short eruption durations (~1 day to 15 yr), the magmas 56 often undergo limited evolution by fractional crystallization and 57 assimilation. In spite of their relative abundance, the hazards posed by 58 cinder cone eruptions are commonly underestimated, and the 59 eruption processes that form cinder cones are not well understood. 60 Most cinder cone eruptions are brief (<1 yr; Wood, 1980) and the 61 magmas do not evolve over the course of the eruption. Longer lived 62 eruptions, like Jorullo and Parícutin in Mexico (15 and 9 yr, 63

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respectively) are more complex; at both volcanoes, magmas evolved 64 65 progressively during the eruptions, and such evolution has been explained by combinations of fractional crystallization and crustal 66 67 assimilation (McBirney et al., 1987; Luhr, 2001; Rubin et al., 2004). Recently, fractionation and assimilation in the lower crust in a "deep 68 69 crustal hot zone" (Annen et al., 2006) or "MASH zone" (Hildreth and 70 Moorbath, 1988) has been invoked to explain intermediate magma 71 compositions erupted from long-lived stratovolcanoes. However, in 72monogenetic systems, where do fractionation and assimilation occur? 73A recent study of an eroded cinder cone has shown a network of dikes and sills feeding the volcano just beneath the surface (Valentine and 74Krogh, 2006). Such a network provides possible shallow magma 75storage locations where magma could evolve in composition; 76 however, the development and evolution of such plumbing systems 77during the course of an eruption are not known. 78

79 Here we use melt inclusions trapped in forsterite (Fo)-rich olivine $(Fo_{85-91}, where Fo = 100 * Mg/(Mg + Fe))$, combined with measurements 80 81 of melt crystallinity and estimates of crystal residence times and magma storage depths, to characterize changes in the plumbing system during 82 the eruption of Jorullo. Our data constrain the undegassed, near-primary 83 volatile contents of the melt and track the degassing, crystallization and 84 85 compositional evolution of the magma before and during the eruption, 86 thereby providing new insight into cinder cone eruption processes. Our results show that cinder cones, in spite of their relatively small size, are 87 capable of evolving both compositionally and structurally during 88 eruption. In longer-lived eruptions like Jorullo (15 yr), the plumbing 89 systems may be more complex than a simple feeder-dike. 90

91 **2. Geologic setting and eruption history**

Jorullo is located in the Michoacán–Guanajuato Volcanic Field (MGVF), in the central part of the subduction-related Trans-Mexican Volcanic Belt (Fig. 1). Volcanism in the MGVF is due to subduction of 94 the Cocos plate beneath the North American plate off the southwest 95 coast of Mexico. The MGVF contains nearly 1000 cinder cones, many of 96 them Holocene in age (Hasenaka and Carmichael, 1985). Jorullo is 97 located along the volcanic front in the MGVF, roughly 80–90 km above 98 the Cocos plate (Pardo and Suárez, 1995). The most recent activity in 99 the MGVF was the eruption of the cinder cone Parícutin from 1943– 100 1952, 84 km NW of Jorullo. 101

Jorullo erupted from 1759 until 1774, producing a thick blanket of 102 tephra and ash as well as numerous lava flows (Luhr and Carmichael, 103 1985). Early accounts indicate that the eruption began violently, with 104 explosive activity that persisted for many years (Gadow, 1930). These 105 accounts, combined with the abundance of fine ash in the tephra 106 blanket, suggest that like Parícutin, Jorullo exhibited violent Strombo- 107 lian activity, which is characterized by simultaneous explosive 108 eruptions producing large amounts of fine ash and effusion of lava 109 from the base of the cone (Pioli and Cashman, 2006). The widespread 110 lava flows at Iorullo were the focus of an important study by Luhr and 111 Carmichael (1985) that described the eruption and the compositional 112 evolution of the lavas over time. They noted that the earliest lavas were 113 primitive basalt, with 9.3 wt.% MgO and phenocrysts of magnesian 114 olivine (Fo_{86-90}), and that as the eruption progressed, the lavas evolved 115 to basaltic andesite compositions (Luhr and Carmichael, 1985). 116

3. Sample description and analytical procedures

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We present results from two sections excavated through the 118 tephra deposit, located ~1 km from the vent to the south and 119 southeast, respectively (Fig. 1). The tephra sections expose the earliest 120 erupted explosive material (in contact with the underlying paleosol) 121 and are composed of bedded ash and lapilli layers that provide a time 122 sequence of the eruption. The tephra also contains abundant, loose 123



Fig. 1. Map of Jorullo and surrounding area, modified from Luhr and Carmichael (1985). Squares mark locations of the two tephra sections used in this study; the 2004 section contains the 'Early' (in contact with the paleosol) and 'Middle' samples, and the more complete 2006 section contains the 'Late' sample. Inset shows plate tectonic boundaries associated with the Trans-Mexican Volcanic Belt, with major volcanoes shown as dots.

olivine phenocrysts throughout the section. Samples for this study
were chosen from three representative tephra layers – the basal
(in contact with the paleosol), middle, and upper parts of the section –
that will be referred to as the 'early', 'middle', and 'late' samples,
respectively.

We analyzed olivine-hosted melt inclusions to obtain pre-eruptive 129volatile and melt compositions. Loose olivine crystals from each of the 130three layers were separated, cleaned in HBF4, and examined in 131 132immersion oil (refractive index 1.678). Most olivine crystals are 133 euhedral, but subhedral and skeletal crystals are also present. Melt inclusions identified as suitable (those that are fully enclosed by host 134crystal, are bubble-poor, and appear to be sealed) were prepared as 135doubly polished wafers and were analyzed for H₂O and CO₂ (FTIR), 136major elements, S, and Cl (electron microprobe), and trace elements 137(laser ablation ICP-MS). Melt inclusion compositions were corrected 138 for post-entrapment crystallization of olivine and Fe-loss; corrected 139 values are shown in Table 1 and uncorrected values are shown in 140 Supplementary Table 1 (for details on the correction method and error 141 propagation see Supplementary material). All data shown in figures 142and discussed in the text are corrected values. Bulk tephra samples 143 from the three layers were analyzed by XRF at Washington State 144 University for major and trace elements (Supplementary data Table 2). 145 146 We also analyzed the compositions of olivine phenocrysts (Table 1) and tephra groundmass glass (Supplementary data Table 3) through-147

out the tephra stratigraphy. Major elements, S, and Cl were measured 148 with a Cameca SX-100 electron microprobe at the University of 149 Oregon using a 15 kV accelerating voltage, 10 nA beam current, and a 150 beam diameter of 10 μ m. A combination of glass and mineral stan- 151 dards was used. Trace element concentrations in the melt inclusions 152 were measured by laser ablation ICP-MS at Oregon State University 153 (Supplementary data Table 4). Details of the technique are found in 154 Kent et al. (2004).

Water and CO₂ concentrations in melt inclusions were analyzed by 156 Fourier Transform Infrared Spectroscopy (FTIR) at the University of 157 Oregon. Concentrations of H₂O and CO₂ were calculated using Beer's law: 158 $c=MA/\rho d\varepsilon$, where M is the molecular weight of H₂O or CO₂, A is the 159 measured absorbance of the band of interest, ρ is the room temperature 160 density of basaltic glass, *d* is the thickness of the melt inclusion and ε is 161 the molar absorption coefficient. In most samples, water concentrations 162 were calculated using the total OH peak at 3550 cm⁻¹ and an absorption 163 coefficient of 63±3 L/mol cm (P. Dobson et al., unpublished data, cited by 164 Dixon et al., 1995). In some instances, however, total H₂O was calculated 165 using an average of the molecular H_2O peaks at 1630 cm⁻¹ and 5200 cm⁻¹ 166 and the OH⁻ peak at 4500 cm⁻¹. In these cases absorption coefficients 167 were calculated based on major element compositions (Dixon et al., 168 1995) and are as follows: 1630 cm⁻¹=26.4 L/mol cm, 5200 cm⁻¹= 169 0.66 L/mol cm, 4500 cm⁻¹ = 0.56 L/mol cm. CO₂ was calculated using 170the carbonate peaks at 1515 and 1435 cm⁻¹; an absorption coefficient 171

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Group inclusion	Early 1	Early 12_1	Early 12_2	Early 12_3	Early 13_1	Early 13_2	Early 14	Early 17_1	Early 17_2	Early 17_A	Early 20	Early 23	Early 24	Early 28	Early 29_1	Early 29_3
SiO ₂	51.15	51.54	52.06	51.58	50.60	50.11	50.93	52.90	51.32	51.19	51.22	50.63	3 49.89	51.57	51.01	50.26
TiO ₂	0.79	0.68	0.63	0.75	0.88	0.75	0.67	0.74	0.80	0.74	0.88	0.75	0.73	0.65	0.76	0.71
Al_2O_3	15.66	16.77	17.25	16.96	18.90	19.46	17.15	16.44	17.21	17.46	16.74	17.15	17.87	16.63	17.69	17.55
FeO ^T	7.84	7.38	7.30	7.36	7.65	7.26	7.46	7.47	7.52	7.41	7.55	7.56	7.62	7.48	7.65	7.57
MnO	0.15	0.13	0.06	0.08	0.08	0.21	0.02	0.10	0.10	0.04	0.15	0.07	0.05	0.10	0.14	0.09
MgO	10.51	9.23	9.00	9.44	7.75	7.43	9.99	9.29	9.64	9.25	9.91	10.03	10.46	9.82	8.69	9.56
CaO	9.20	9.34	8.57	8.91	8.92	9.21	9.01	8.28	8.55	8.93	9.14	8.96	8.60	8.55	8.92	9.21
Na ₂ O	3.74	4.13	4.20	4.04	4.26	4.59	3.89	4.01	4.04	4.17	3.60	4.00	3.92	4.25	4.21	4.21
K ₂ O	0.77	0.65	0.77	0.73	0.81	0.83	0.70	0.65	0.67	0.68	0.61	0.66	0.72	0.84	0.77	0.66
P_2O_5	0.18	0.15	0.16	0.15	0.16	0.16	0.17	0.13	0.15	0.14	0.19	0.19	0.14	0.12	0.16	0.16
H ₂ O	0.9	4.7	2.5	4.2	3.4	n.a.	1.2	5.4	2.6	n.a.	1.5	3.8	2.2	1.0	2.9	3.4
CO ₂ ppm	118	243	207	915	-	n.a.	307	805	692	n.a.	594	988	-	-	358	556
S ppm	1149	1195	1379	1751	1591	1560	1718	1751	1665	2066	1847	1882	1776	688	1737	1644
Cl ppm	1103	1179	1197	1239	1285	1406	1381	1239	1308	1286	1267	1196	1185	1105	1176	1206
Total *	99.17	97.05	97.18	96.28	97.07	98.22	97.50	96.18	96.14	95.42	95.34	94.10	95.88	96.99	96.36	96.72
P (bars)	343	2542	1064	3445	1152	n.a.	808	4027	2080	n.a.	1481	3444	515	101	1668	2309
Olivine (Fo%)	91.1	90.3	89.9	90.4	88.4	88.4	90.5	90.1	90.3	90.1	90.5	90.7	90.6	90.0	89.3	90.2
%PEC **	14.2	6.9	12.4	10.6	3.5	4.4	17.9	9.1	12.5	9.7	11.6	10	20.6	18	8.5	11
Group inclusion	Middle 4	Midd	le 5 🛛 🕅	Middle 6	Middle 7	7_1 M	ddle 10	Late 1	Late 3_1	Late 3_	2 Late	4 Late	5 Lat	e p1	Late p2	Late p3a
SiO ₂	52.92	52.99) 5	50.15	50.24	51	.70	51.34	52.54	55.21	53.66	51.9	5 50.	56	54.13	55.14
TiO ₂	0.76	0.88	().90	0.89	0.3	76	0.98	0.95	1.12	0.88	0.84	1.00	6 (0.98	0.91
Al ₂ O ₃	17.42	18.09) 2	20.14	19.69	20	.32	19.83	19.07	18.02	18.01	20.4	2 20.	25	18.59	17.87
FeO ^T	7.48	6.75	6	6.79	7.07	6.9	92	6.76	7.07	6.69	7.47	6.68	6.8	9 (6.73	6.76
MnO	0.12	0.11	(0.10	0.12	0.0)7	0.08	0.15	0.03	0.06	0.10	0.0	8 (0.06	0.04
MgO	7.26	6.53	5	5.97	6.93	6.8	30	6.06	5.83	5.53	5.91	5.15	6.15	5 (6.05	6.06
CaO	9.05	8.87	1	0.61	9.70	7.2	28	9.71	8.96	8.29	8.13	9.55	9.5	0	8.26	8.72
Na ₂ O	4.02	4.45	4	1.32	4.30	5.0)6	4.29	4.34	4.07	4.66	4.24	4.5) · (4.28	3.72
K ₂ O	0.82	1.15	().83	0.84	0.9	93	0.74	0.85	0.85	1.02	0.83	0.7	2	0.75	0.60
P ₂ O ₅	0.15	0.17	().20	0.21	0.1	7	0.22	0.23	0.20	0.20	0.24	0.19) (0.17	0.17
H ₂ O	1.4 0.5 1.1		1.1	0.8	3	0.5	0.8	0.8	0.8	1.0	n.a.	1	n.a.	n.a.		
CO ₂ ppm	-	-	-	-	-	-		-	-	-	-	-	n.a.	I	n.a.	n.a.
S ppm	1407 1299 2040		1786	17	81	1442	1143	74	918	1490) 210	6	1477	1479		
Cl ppm	910 1247 1094		1127	14	19	1239	1162	1020	994	1065	5 129	8	1206	936		
Total*	99.47 102.86 97.24		7.24	99.93	10	0.11	100.10	100.01	99.25	99.39	98.5	1 100	.02	100.00	99.96	
P (bars)	186	24	1	15	121	62		26	64	65	68	87	n.a.	1	n.a.	n.a.
Olivine (Fo%)	88.2	88.0	8	36.6	88.4	87	.3	86.7	85.6	85.6	84.9	85.0	86.	7	86.2	86.8
%PEC**	4.8	6.7	2	2.0	3.8	11	2	4.5	5.9	6.4	8.1	1.6	4.1		11.8	2.3

Total* = sum of all oxides plus S and Cl in original (uncorrected) microprobe analyses, %PEC** = wt.% post-entrapment crystallization. All melt inclusion data were corrected for postentrapment crystallization (%PEC) of olivine (Sobelev and Chaussidon, 1996) and diffusive loss of Fe (Danyushevsky et al., 2000; see Supplementary material). Major element oxides reported are normalized to 100% on a volatile-free basis. The average standard deviation (absolute) based on multiple (2–4 points) analyses is as follows: SiO₂: 0.27, Al₂O₃: 0.20, FeO: 0.16, MgO: 0.22, CaO: 0.36, Na₂O: 0.27, K₂O: 0.03, TiO₂: 0.10, MnO: 0.05, P₂O₅: 0.03, S: 0.009, Cl: 0.006.

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Fig. 2. Major elements vs. MgO for Jorullo melt inclusions, tephra groundmass glass, bulk tephra, and lava whole-rock samples (lava analyses are from Luhr and Carmichael, 1985). In both figures, MELTS calculations (see Fig. 8) indicate that olivine-only crystallization explains much of the range in MgO, K₂O, and CaO whereas some inclusions record plagioclase±clinopyroxene crystallization. The MELTS calculations were done to simulate a probable ascent path (varying both pressure and temperature), the solid line shows an ascent path with greater cooling than the path of the dotted line. In contrast, the bulk melt evolution, as shown by the lava trend, corresponds to fractionation of 17.3 wt.% amphibole+6.7 wt.% olivine (thick dashed line), based on major element modeling. Also shown are the average standard deviations based on multiple analyses per sample.

of 296 L/mol cm was calculated following Dixon and Pan (1995). Based on uncertainties in thickness measurements and absorbance values, average 1 standard deviation uncertainty in H₂O is \pm 0.2 wt.% and in CO₂ \pm 80 ppm. The background subtraction procedure that we used for the carbonate peaks is described in Roberge et al. (2005).

We measured compositional profiles across zoned olivine crystals by electron microprobe. We calculated residence times for the olivine crystals predicated on the assumption that the zoning was produced by a magma mixing event (discussed in Section 5.5), and followed the methods of Costa and Chakraborty (2004). We used the simple, 1-*D* diffusion equation:

$$C = C_1 + (C_o - C_1) * \operatorname{erf}\left[x / \left(2\sqrt{D^*t}\right)\right]$$
(1)

where C is the concentration, x is distance, D is the diffusion coefficient, and t is time.

We calculated diffusion coefficients, *D*, for Fe–Mg exchange using the equation from Jurewicz and Watson (1988):

$$D_{\rm Fe-Mg} = 8 \times 10^{-7} \exp(-29708/T) m^2 s^{-1}$$
⁽²⁾

where *D* is diffusion parallel to the *c*-axis, and we assumed that $D_a = D_b = 6D_c$ (Dohmen et al., 2003). Using Eq. (1) we fit a modeled diffusion profile to our data, and calculated the time required to produce that profile. Our calculations were based on compositional profiles along the *a*- and *b*-axes of olivine crystals, and we used eruption temperatures calculated using the method of Sugawara (2000), which generally agree with eruption temperatures calculated using

MELTS and pMELTS (Ghiorso and Sack, 1995; Asimow and Ghiorso, 196 1998; Ghiorso et al., 2002). 197

To track changes in crystallinity of the melts throughout the 198 eruption we took images of the groundmass from the three 199 representative tephra layers. Images were collected using an FEI 200 Quanta scanning electron microscope (SEM) at the University of 201 Oregon. These images were then analyzed using SCION image analysis 202 software to calculate the groundmass crystallinity and the size (crystal 203 area) of groundmass microlites for each sample. 204

4.1. Melt inclusion and groundmass compositions

205 206

Jorullo lavas evolved from initially primitive basalt to basaltic 207 andesite over the course of the ~15 yr eruption (Luhr and Carmichael, 208 1985). Our corrected melt inclusion compositions show a similar 209 evolutionary pattern (Table 1), though there are some important 210 compositional differences between melt inclusions and lavas. The early 211 melt inclusions have high MgO (7.5–10.5 wt.%) and low K₂O (0.6–212 0.8 wt.%), suggesting that early olivine crystals trapped primitive melts 213 that had undergone little to no differentiation from a parental magma 214 (Fig. 2). The later-erupted melt inclusions sampled progressively more 215 evolved melt compositions, with decreasing MgO (5–7.4 wt.%) and 216



Fig. 3. Trace element analyses for melt inclusions (laser ablation ICP-MS) and bulk tephra (ICP-MS). Fig. 3a shows Ba vs. La and Fig. 3b shows Zr vs. Y. Both figures show a solid line representing 40% fractionation (an upper limit of crystallization, as suggested by Fig. 9) and an analysis of the La Huacana granite bedrock (Luhr and Carmichael, 1985). Light gray bands indicate mixing between the Jorullo melts and the granite, suggesting a role for assimilation in the later melts.

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increasing $K_2O(0.8-1.1 \text{ wt.}\%)$. Groundmass glasses have low MgO (3.7-4.3 wt.%) and higher $K_2O(1.3-1.6 \text{ wt.}\%)$, indicating that extensive crystallization occurred after melt inclusion entrapment. Importantly, the lavas show increases in K_2O and decreases in CaO with decreasing MgO that differ from many of the melt inclusions. These deviations imply differences in magmatic processes for lavas and melt inclusions, and the significance of this will be discussed in Section 5.3.

Melt inclusion incompatible trace element concentrations also generally increase from the early to late stages of the eruption (Fig. 3). While some elements show steady increases from early to late in the eruption, several elements seem to be anomalously enriched in the late melt inclusions. Similar enrichments were seen in some trace elements in the late-stage lavas (Luhr and Carmichael, 1985).

The bulk tephra analyses from the early, middle and late tephra 230layers are all less evolved than the earliest lavas (Fig. 2), suggesting 231 that the preserved tephra sequence that we sampled may have been 232 erupted prior to most, if not all, of the lava flows. This interpretation is 233 supported by descriptions of the eruption that suggest that the early 234phases of activity were mainly explosive and that lava flows may not 235have issued from the cone until the eruption was in its fifth year 236 (Gadow, 1930). An alternative possibility is that the proximal tephra 237deposits that we sampled were enriched in dense olivine relative to 238 239the lower density, vesicular pyroclasts ejected during eruption and 240 fallout of the tephra.

241 4.2. Olivine compositions and zoning

Comparison of olivine compositions in the tephra and lava supports our interpretation that the explosive eruptions largely preceded effusive activity, and shows that the olivine forsterite content decreased throughout the eruption. Fig. 4 shows probability



Fig. 4. Probability density curves for olivine core compositions from tephra (lower) and lava (upper; Luhr and Carmichael, 1985), where E = Early, M = Middle, L = Late (for olivine from both tephra and lava), and LS = latest (lava). These plots illustrate the general decrease in olivine core Fo-content throughout the eruption. Early and middle tephra samples appear to correspond to an earlier phase of the eruption based on high olivine Fo-content and the narrow range of olivine compositions, and the primitive composition of bulk tephra. Observations of the eruption that describe explosive activity only during the first few years of the eruption (Gadow, 1930) and a thinning of tephra onto the earliest lava flows support this interpretation. The early phase of erupted lava appears to correlate roughly with our middle to late tephra.

density curves for olivine core compositions from the lava flows (Luhr 246 and Carmichael, 1985) and the tephra throughout the eruption. 247 Olivine from the early and middle tephra are overall more forsterite- 248 rich than olivine from the early and middle lava flows, suggested they 249 were erupted prior to effusion of the lavas. The olivine cores exhibit an 250 increasing compositional range from the early tephra (Fo₈₈₋₉₁) to the 251 late tephra (Fo₈₈₋₈₈) to the latest lava flows (Fo₇₃₋₈₇). 252

Olivine crystals from both the tephra and lava flows are normally 253 zoned (Fig. 5). Zoned rims range from abrupt in the early tephra (Fig. 5a) 254 to more gradual in the late tephra (Fig. 5b), suggesting that crystals may 255 have resided in more evolved magmas for varying lengths of time. Using 256 transects across these zoned crystals and the equations described earlier, 257 we have modeled the zoning profiles (solid lines, Figs. 5a, b) and the 258 corresponding residence times for olivine from the early, middle and late 259 stages of the eruption. We calculate a range of olivine residence times, 260 from ~10–200 days for the early-erupted olivine to ~80–1300 days for 261 the late-erupted olivine (Fig. 5c). Implications for these residence times 262 are discussed in Section 5.5.

Melt inclusions preserved in Jorullo olivine trap some of the 265 highest volatile contents yet recorded in primitive arc magmas and 266 suggest a complex and evolving degassing history throughout the 267 eruption (Fig. 6). Inclusions in early erupted olivine record a wide 268 range of volatile contents and trapped both relatively undegassed 269 melts, with up to 5.3 wt.% H₂O and 1000 ppm CO₂, and degassed melts, 270 with low H₂O and CO₂ below detection (<50 ppm). This diversity in 271 H₂O and CO₂ concentrations indicates that olivine crystallized over a 272 wide range of depths, corresponding to entrapment pressures of 10 to 273 ~400 MPa. The later-erupted inclusions are much more uniform in 274 volatile concentrations, with all melt inclusions recording CO₂ below 275 detection and consistently low H₂O (0.2–1.4 wt.% in middle samples; 276 0.4-1.0 wt.% in late samples) that indicate olivine crystallization only 277 at very low pressures (<20 MPa). In contrast, melt inclusion S and Cl $_{\rm 278}$ contents do not vary significantly during the eruption (S mostly 1200- 279 1800 ppm; Cl 1000–1400 ppm, Table 1) and do not correlate with H₂O. 280

4.4. Groundmass crystallinity

The tephra groundmass contains abundant microlites of plagio- 282 clase, as well as minor olivine and clinopyroxene (Fig. 7a, b). Mea- 283 surements of microlite abundance in the groundmass from the 284 three tephra layers show a slight increase in crystallinity from early 285 (41%±1%, average±1 s.e.) to late (47%±2%) in the eruption (Fig. 7c). 286 Additionally, there are notable increases in microlite size (area), as 287 seen visually in the comparison of the early groundmass (Fig. 7a) and 288 late groundmass (Fig. 7b). The average size of olivine and clinopyr- 289 oxene crystals increases from $60\pm13 \,\mu\text{m}^2$ (early) to $145\pm13 \,\mu\text{m}^2$ (late), 290 and the average size of plagioclase laths increases from $124\pm7 \,\mu\text{m}^2$ 291 (early) to $180\pm20 \,\mu\text{m}^2$ (late) (Fig. 7d).

5.1. Degassing processes

The scatter in the melt inclusion H_2O and CO_2 concentrations 295 shown in Fig. 6 reflects a complex magma degassing history. 296 Calculated degassing paths (solid lines, Fig. 6) show that some of the 297 variation in H_2O and CO_2 can be explained by either open-system 298 degassing, where CO_2 degasses almost entirely with little loss of H_2O , 299 or closed-system degassing, in which the ascending melt and 300 exsolving gas remain in equilibrium. However, many inclusions have 301 elevated CO_2 for a given H_2O content that cannot be explained by 302 closed-system degassing models. Similar scatter has been seen in 303 other magmatic systems (e.g., Rust et al., 2004; Atlas et al., 2006; 304

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Fig. 5. Olivine zoning and residence time calculations. Fig. 5a and b show representative electron microprobe transects from core-to-rim for early and late olivine crystals, respectively. Also shown are modeled diffusion profiles (thick solid lines). Fig. 5c shows the calculated residence times for olivine from the early, middle and late stages of the tephra section. Fig. 5d illustrates the positive relationship between olivine residence time and olivine core composition from the late olivine samples. This suggests that earlier, less evolved melts crystallized olivine that then resided for potentially long periods of time in more evolved melts. Standard deviations were calculated based on multiple transects and models, but error bars are smaller than the symbol size.

Spilliaert et al., 2006). The low H₂O, high CO₂ inclusions could result from several processes, including disequilibrium degassing of melts before entrapment (Gonnermann and Manga, 2005) or post-entrapment diffusive loss of H_2 of H_2O through the melt inclusion host 308 crystals. The former process is difficult to evaluate because of the lack 309 of data on H_2O and CO_2 diffusivities in hydrous basaltic melts (Baker 310



Fig. 6. Melt inclusion CO₂ vs. H₂O for early (diamonds), middle (squares), and late (triangles) samples. Also shown are calculated vapor saturation isobars, open and closed-system degassing paths, and vapor isopleths for 40, 60 and 80% CO₂, all calculated using VolatileCalc (Newman and Lowenstern, 2002). For many inclusions, the pressures based on dissolved H₂O and CO₂ are minimum values because the inclusions contain a small shrinkage vapor bubble (formed post-entrapment) that contains some additional CO₂. However, most low H₂O, low CO₂ inclusions contain no such shrinkage bubble. The average standard deviation was calculated based on multiple analyses per inclusion.

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Fig. 7. Groundmass crystallinity from SEM image analyses. a and b show images of groundmass tephra from the early and late tephra samples, respectively. c shows the average crystallinity for early, middle and late samples based on SCION image analysis estimates. d shows an increase in size of groundmass plagioclase from early to late in the eruption, where size is the area of a groundmass crystal as calculated using SCION image analysis software. Error bars show ±1 standard error based on analysis of multiple images for each sample.

et al., 2005), but it is probably more effective in low temperature, 311 silicic magmas. Loss of water by H₂ diffusion is probably limited to 312 ≤1 wt.% H₂O by redox effects (Danyushevsky et al., 2002), and thus 313 could not explain all of the scatter we see in our data. Loss of water by 314molecular H₂O diffusion through the olivine host (Portnyagin and 315 Almeev, 2007) is not limited by redox reactions. However, the loss of 316 2.5 to 4 wt.% H₂O that would be required to account for our data 317 (assuming all high CO₂ inclusions started along the closed-system 318 degassing curve in Fig. 6) would cause considerable crystallization of 319 olivine, plagioclase, and clinopyroxene inside the inclusions and 320 formation of a substantial shrinkage vapor bubble. None of these 321 features are observed in the high CO₂, low H₂O inclusions, leading us 322 to conclude that significant H₂O diffusive loss through the host olivine 323 324 has not occurred. Additionally, we see no correlation between diffusive Fe-loss from the melt inclusions (see Supplementary 325326 material) and the deviation of H₂O contents from the closed-system degassing curve in Fig. 6. 327

An alternative possibility is that the scatter in Fig. 6 may be the 328 result of melts equilibrating with more CO₂-rich vapor percolating 329 through the system from below, where it is released by magma 330 **O2**331 degassing deeper in the system (Rust et al., 2004; Spilliaert, 2006). Vapors with 40–80 mol% CO₂ fit most of the scatter in our data (CO₂ 332 vapor isopleths, dashed lines, Fig. 6). This gas fluxing hypothesis 333 requires that melts have initially high CO_2 contents (≥ 0.7 wt.%) in 334order to create such CO₂-rich gases during ascent in the deep crust. 335 Although such high values have not been found in melt inclusions, an 336 analysis based on arc volcanic CO₂ fluxes suggests that such high 337 values are common in mafic arc magmas (Wallace, 2005). The gas 338 fluxing effect may be enhanced by repeated cycles of decompression 339 and repressurization within the conduit (Rust et al., 2007). 340

The melt inclusions record evidence for loss of H₂O and CO₂ by degassing, but there is no indication of S and Cl degassing throughout the eruption, probably because these components have higher solubility. Lack of Cl degassing has been observed previously (e.g., Sisson and Layne, 1993), but the lack of S degassing is more peculiar and may result from relatively high oxygen fugacity and presence of 346 S primarily as sulfate. 347

5.2. Crystallization during ascent and degassing

The primitive composition of the early Jorullo melts combined with 349 initially high magmatic volatiles, gas fluxing, and degassing caused 350



Fig. 8. Phase diagram for early Jorullo melt composition (10.5 wt.% MgO) constructed using MELTS (Ghiorso and Sack, 1995; Asimow and Ghiorso, 1998) and pMELTS (Ghiorso et al., 2002). Circle with cross indicates hypothetical starting melt based on the highest melt inclusion entrapment pressure of 400 MPa with 5.3 wt.% H₂O. The adiabatic ascent path was calculated using MELTS and pMELTS and includes the effects of crystallization, gas exsolution, and gas expansion; the widening at low pressures reflects variability between the two models. Amphibole liquidus region is based on experimental studies by Holloway (1973), Holloway and Ford (1975), Moore and Carmichael (1998), Grove et al. (2003), and Nicholis and Rutherford (2004).

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olivine-only crystallization over the wide range of depths indicated by 351 352 our melt inclusion data. These inferences are supported by a phase diagram for early Jorullo melts constructed using MELTS and pMELTS 353 354(Fig. 8). The high MgO and H₂O contents of the early Jorullo melts produce a large field of olivine-only crystallization. Melt inclusion data 355 demonstrate that olivine crystals formed at a maximum pressure of 356 400 MPa; however, at this pressure the melts are H₂O-undersaturated. 357 Deep crystallization would have been facilitated by CO₂-rich vapor 358 359 fluxing through the system, as suggested by some of the melt inclusion 360 data (Fig. 6). Fluxing of CO₂-rich gas would remove H₂O from the melt, 361 even though the melts were H₂O undersaturated. Thus the melts would have been below their relevant H₂O undersaturated liquidii (Fig. 8), 362 forcing small quantities of olivine to crystallize during ascent. Larger 363 364 amounts of olivine would have crystallized once the melts crossed the H₂O saturated liquidus (at ~150-200 MPa). 365

The phase relations (Fig. 8) are also consistent with the com-366 plicated degassing history shown in the plot of CO₂ vs. H₂O (Fig. 6). In 367 this plot there are essentially two groups of inclusions; the first group 368 are those that roughly fit either open or closed-system degassing 369 paths, in other words, melts that have degassed all of their CO₂ prior to 370 olivine crystallization. These crystals would have grown along or 371 below the H₂O-saturated olivine-in curve at pressures commencing at 372 373 ~150 MPa, as recorded by the entrapment pressure of the CO₂-poor 374 melt inclusion with the highest H_2O (Fig. 6). The second group of inclusions are those that cannot be explained by either closed or open 375 system degassing, and instead require fluxing of CO₂-rich vapor 376 through the system. According to our melt inclusion data, nearly all 377 378 inclusions trapped at pressures >200 MPa require interaction with CO₂-rich vapor; these pressures correspond to the region above the 379 H₂O-saturated olivine-in curve where the ability to crystallize olivine 380 381 depends on addition of CO_2 and subsequent loss of H_2O from the melt.

382 The adiabatic ascent path in Fig. 8 shows that olivine would be the 383 only crystallizing phase over a large range of pressures (~400-30 MPa), with plagioclase joining olivine-only at low pressures 384 (<30 MPa), followed by near-surface crystallization of clinopyroxene. 385 This predicted ascent and crystallization path agrees with the melt 386 inclusion data, the modal abundance of olivine in the tephra, and the 387 388 presence of late-stage plagioclase and minor clinopyroxene as groundmass crystals in the quenched tephra glass (Fig. 7a, b). 389

The later-erupted melt inclusions indicate that crystallization 390 moved to shallow levels as the eruption progressed. The lower MgO 391 392 content of the later melts would have decreased the olivine-only field, such that olivine could only have crystallized at pressures ≤100 MPa 393 (based on an eruption temperature of ~1150 °C and the phase diagram 394 395 of Moore and Carmichael, 1998). However, melt inclusion entrapment pressures later in the eruption vary only from 3 to 19 MPa, suggesting 396 397 that either deeper crystallizing olivine fractionated out of the melt and was not erupted, or, that olivine preferentially crystallized only 398 shallowly in the plumbing system. This shallow crystallization further 399 suggests that the CO₂ fluxing that enabled deep olivine crystallization 400 early had shut off later in the eruption. Additionally, some later-erupted 401 402 melt inclusions show the compositional effects of plagioclase +/-403 clinopyroxene crystallization (see Section 5.3). These phases would also have crystallized at low pressures (<40 MPa), lending further support to 404 the development of a shallow region of melt crystallization. 405

Our data show that the crystallization recorded by melt inclusions is 406 407driven by H₂O loss during ascent. At higher pressures (200-400 MPa) loss of H₂O is likely caused by gas fluxing, but at lower pressures, CO₂-depleted 408 melts lose H₂O by direct exsolution of H₂O-rich vapor. Both processes 409 cause melts to degas H₂O and thus can result in crystallization. Variations 410 in K₂O, which is incompatible during crystallization, are consistent with 411 a model of degassing-induced crystallization (Fig. 9). Based on the K₂O 412 content of the highest pressure inclusion (0.66 wt.% K₂O), increases in K₂O 413 with decreasing pressure for all early erupted melt inclusions are con-414 sistent with up to 14% crystallization. Later in the eruption, low pres-415 416 sure degassing resulted in extensive shallow crystallization of olivine,



Fig. 9. Pressure (MPa) vs. K₂O for melt inclusions and groundmass glasses. Pressures for melt inclusions were calculated based on dissolved CO₂ and H₂O (Fig. 6) using the VolatileCalc solubility model (Newman and Lowenstern, 2002). Percent total crystallization required to relate various compositions to the parental melt is calculated assuming that K₂O is perfectly incompatible. Also shown is the calculated MELTS ascent path, which predicts 48% total crystallization and correlates very well with the data (gray circle). Error bars show ± 1 standard deviation based on multiple analyses. Where not shown, error is smaller than symbol size.

plagioclase, and minor clinopyroxene, producing K_2O increases that 417 require up to 29–36% crystallization prior to melt inclusion entrapment 418 (Fig. 9). 419

Whereas the late inclusions were trapped shallowly (3–19 MPa), 420 crystallization of the groundmass in the tephra clasts occurred shallower 421 still, likely during rise in the upper conduit and during eruption. 422 Groundmass glass analyses for the three tephra layers indicate that 423 significant crystallization of microlites occurred after the melt inclusions 424 were trapped (Fig. 9). The K₂O in tephra groundmass glass increases 425 steadily from the early (1.3 wt.%) to late (1.6 wt.%) samples, correspond- 426 ing to approximately 50% total crystallization (early) and 59% total 427 crystallization (late) of initially primitive melt. These values suggest that 428 \sim 25% crystallization occurred between the last melt inclusion trapped 429 and eruption and quenching of the pyroclasts, and that much of this 430 extensive crystallization was shallow (<10 MPa).

Additionally, the temporal increase in both crystallinity and crystal 432 size measured in groundmass glass images (Fig. 7c, d) suggests an 433 increased time scale for groundmass crystallization prior to eruption. 434 This could have been achieved by either decreasing ascent rate in the 435 latter stages of the eruption or storage of later melts en route to the 436 surface. Shallow storage of magma at the base of a growing cinder 437 cone has been suggested elsewhere (Krauskopf, 1948; Cervantes and 438 Wallace, 2003; Pioli and Cashman, 2006) to explain both lava effusion 439 from the base of cinder cones and shallow growth of olivine crystals. 440

5.3. Lava vs. tephra – the role of deep crustal fractionation 441

The melt compositions trapped in olivine during the eruption of 442 Volcán Jorullo record the importance of olivine crystallization through- 443 out the eruption, with the addition of shallow plagioclase +/- 444 clinopyroxene later in the eruption. Olivine crystallization played a key 445 role in changing the liquid composition, as evidenced by the abundance 446 of olivine in the tephra and lava flows and the decrease in melt inclusion 447 MgO contents as the eruption progressed (Fig. 2). Much of the melt 448 inclusion data can be explained by ~13 wt.% olivine fractionation (solid 449 line in Fig. 2). However, lower CaO contents in some of the later 450 inclusions and the groundmass glass require additional crystallization of 451 plagioclase +/- clinopyroxene (solid and dotted lines, Fig. 2b). 452

Whereas the melt inclusion data record crystallization of olivine \pm 453 plagioclase \pm clinopyroxene in the upper crust caused by ascent and 454 degassing, the bulk lava compositions (gray field, Fig. 2) differ from the 455 melt inclusions, suggesting a different fractionation history. Unlike the 456

melt inclusions, the lavas gradually increase in K₂O and decrease in 457 458 CaO with decreasing MgO beginning early in the eruption. As no combination of the main erupted phenocrysts (olivine±plagioclase) 459 460 can generate this trend, Luhr and Carmichael (1985) explained this evolution by high pressure fractionation of clinopyroxene, olivine, and 461 plagioclase. However, they noted that this explanation was proble-462 matic because of the absence of clinopyroxene phenocrysts in most 463 lavas, and abundances of a number of incompatible trace elements 464 465were not satisfactorily accounted for with this fractionation model.

The presence of amphibole phenocrysts in some of the late-stage, 466 467 most evolved, Jorullo lavas (Luhr and Carmichael, 1985) provides 468 strong evidence that at least some portion of the crystallization process 469 occurred at pressures high enough to stabilize amphibole (>500 MPa); 470 we propose that fractionation of amphibole+olivine±clinopyroxene drove the bulk melt evolution at depth. We modeled simple amphi-471bole±olivine fractionation with major element least squares model-472ing using the early and late bulk lava data, the amphibole composition 473from an analyzed phenocryst in a late lava (Luhr and Carmichael, 1985), 474 and Fogo olivine. We found that fractionation of amphibole + olivine has 475a good fit ($\Sigma r^2 = 0.99$) compared to amphibole only ($\Sigma r^2 = 4.66$). Plotted 476 in the major element diagrams of Fig. 2 is the amphibole+olivine 477 fractionation trend (thick dashed line), which corresponds to fractio-478 479 nation of 17.3 wt.% amphibole and 6.7 wt.% olivine. Because there is uncertainty as to whether amphibole would have been stable in the 480 highest temperature melts (e.g., Moore and Carmichael, 1998; Grove 481 et al., 2003; Nicholis and Rutherford, 2004), the earliest stages of 482 fractionation may have involved olivine+clinopyroxene, with amphi-483 484 bole joining the assemblage (or replacing clinopyroxene by reaction relation) at temperatures between 1100-1150 °C. We hypothesize that 485most of the original fractionating crystals were left behind at depth, 486 perhaps in a deep crustal hot zone sill (Annen et al., 2006), and that 487 488 ascending batches of melt then crystallized olivine during degassing. 489 Thus most or all of the crystals in the tephra and lavas were formed 490 during ascent, whereas the bulk composition of melt batches emanating from the deep reservoir became progressively more 491 evolved over time due to deep fractionation. 492

493 5.4. Role of shallow assimilation – trace elements

While it seems that most of the bulk melt evolution can be explained 494 by fractionation of amphibole + olivine ± clinopyroxene at depth, there is 495also evidence for shallow assimilation of granitic bedrock (Rubin et al., 496 2004). Granitic xenoliths were erupted in several of the middle and late 497 lava flows at Jorullo, and in some localities are quite abundant. 498 Disaggregated xenoliths in the lavas and the presence of plagioclase 499 500 and quartz xenocrysts in thin section suggest efficient consumption of 501the granites by the Jorullo magmas. Additionally, some trace elements are far more enriched in the late melt inclusions and bulk lavas (Luhr and 502Carmichael, 1985) than would be expected from simple fractionation. 503Fig. 3 shows trace element data for melt inclusions and bulk tephra. The 504early and middle samples show increases in trace elements (Ba, La, Zr, Y) 505506 that fit well with simple fractionation (solid line). However, the late melt 507inclusions and late bulk tephra show enrichments in some elements (La, Zr, Y) that do not fit with the modeled 40% crystal fractionation. Also 508plotted in these figures are analyses of the La Huacana granite that 509comprises the bedrock around Jorullo and the xenoliths present in the 510511lava flows; mixing between this granite and Jorullo melts can explain the enrichments in the later samples. Furthermore, these data suggest that 512the plumbing system at Jorullo evolved such that efficient assimilation of 513 the shallow granite occurred, lending additional support to the 514 development of a shallow storage region late in the eruption. 515

516 5.5. Olivine residence times

517 Olivine crystals in both the tephra and lava flows (Luhr and 518 Carmichael, 1985) show Fo-rich cores with more evolved rims. Most olivine from the early and middle tephra have broad homogeneous 519 cores and narrow, normally zoned rims. Olivine from the late tephra 520 have more gradual normal zoning. We interpret the narrowness of the 521 zoned rims in early and middle tephra to be the result of magma 522 mixing or entrainment of earlier formed crystals by less Mg-rich melts 523 just before eruption. Additionally, early olivine crystallized at a wide 524 range of depths but were deposited within the same tephra layer, or 525 eruptive unit. This suggests that olivine from various levels of the 526 plumbing system were re-entrained by the evolving melts rising from 527 depth, consistent with the Jorullo bulk melt evolution driven by 528 amphibole+olivine fractionation in the middle to lower crust. 529

The olivine residence time data (Fig. 5c, d) also support the for- 530 mation of a shallow reservoir or storage system as the eruption 531 progressed. The range of timescales for olivine storage increases from 532 early to late in the eruption, suggesting that longer storage of crystals 533 is facilitated late in the eruption by such a reservoir. Additionally, the 534 observed correlation in the late erupted olivine between crystal 535 residence time and olivine core compositions (Fig. 5d) suggests that 536 the olivine with more Fo-rich cores crystallized far earlier in the 537 eruption (with residence times up to 1300 days), and then resided at 538 shallow levels (based on the low entrapment pressures of middle and 539 late melt inclusions) in a more evolved melt. Together, these data 540 require the development of a region for long-term (months to years) 541 crystal and melt storage at shallow levels, and support the idea of deep 542 melt evolution followed by shallow olivine-dominated crystallization. 543

6. Cinder cone plumbing systems

This research gives new insight into the volatile content, crystal- 545 lization processes, and plumbing system evolution of a cinder cone 546 volcano. The eruption of Jorullo is the longest historically recorded 547 cinder cone eruption, during which time the melts evolved. By 548 comparing the bulk lava record (Luhr and Carmichael, 1985) to the 549 melt inclusion record, we have found evidence for a multi-stage 550 crystallization history at Jorullo. The melt inclusions record extensive 551 crystallization in the upper crust driven by ascent, degassing, and gas 552 streaming, whereas the bulk lava and tephra compositions record 553 deeper fractionation processes. Primitive, volatile-rich magma likely 554 rose from the mantle beneath Jorullo, stalled and cooled in the lower 555 crust, where it fractionated amphibole+olivine during the 15-yr 556 eruption. Such "cryptic amphibole fractionation" in the lower arc crust 557 has recently been suggested by Davidson et al. (2007) to be a 558 widespread phenomenon. 559

Whereas this initial stage of fractionation was likely driven by deep 560 cooling, the Jorullo melt inclusions record evidence for crystallization 561 driven by degassing during magma ascent. The early Jorullo melts rose 562 from depth, likely in a complex network of dikes and sills, and 563 crystallized olivine from 16 km to the near-surface (calculated assuming 564 upper crustal density of 2600 kg/m³). Olivine crystallized at pressures of 565 200–400 MPa due to fluxing of CO₂-rich gases through the conduit 566 system, which forced melts to degas H₂O, and thus placed the melts 567 below the relevant H₂O-suturated olivine-in curves. Subsequent 568 induced further crystallization. The later-erupted melt inclusions record 570 a significant change in the plumbing system, with olivine crystallization 571 localized very shallowly beneath the volcano (80–700 m) and the 572 addition of plagioclase ± clinopyroxene as crystallizing phases.

The shallowing of olivine crystallization depths, combined with both 574 the extensive crystallization recorded by the groundmass and the 575 increasing olivine residence times indicated by diffusion profiles, 576 suggest that a shallow reservoir developed as the eruption of Jorullo 577 progressed, facilitating degassing, crystallization, and melt storage. A 578 similar shallow degassing region was suggested by Krauskopf (1948) to 579 explain the extrusion of degassed lava laterally from the base of 580 Parícutin. This model would also hold for Jorullo, as degassed lavas 581 effused from the base of the cone for much of the eruption. Additionally, 582

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degassing, crystallization, and release of latent heat in this shallow 583 584 network would have enabled crustal assimilation of the granitic bedrock, a process seen both at Parícutin (McBirney et al., 1987) and 585 586Jorullo (Rubin et al., 2004). Development of such a reservoir seems applicable to long-lived cinder cones like Jorullo. For example, the 587eruption of Parícutin lasted for nine years and the melts similarly 588 evolved over time, both due to fractionation and assimilation (Wilcox, 5891954; McBirney et al., 1987). Because such reservoirs are capable of 590591storing melts for extended periods of time and may feed the extensive 592lava flows, they may be a common feature of other long-lived cinder 593cone eruptions.

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600 Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.epsl.2008.03.004.

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